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|  | **Monday** | **Tuesday** | **Wednesday** | **Thursday** | **Friday** |
| **Standards****What SPI’s are being covered for the day?** | SPI 3221.2.1SPI 3221.2.4 | SPI 3221.2.1SPI 3221.2.4SPI 3221.2.6SPI 3221.MATH.1SPI 3221.MATH.2 | SPI 3221.1.1SPI 3221.1.2 | SPI 3221.1.1SPI 3221.1.2 | SPI 3221.1.1SPI 3221.1.2 |
| **Objectives****What do the students need to accomplish by the end of the lesson?** | Students will:-Determine the chemicals the create certain physical and chemical changes | Students will:-Show knowledge of previously learned standards and objectives by scoring at least an 80% on a unit test | Students will:-Compare and contrast the major points of the atomic theory and the modern atomic theory | Students will:-Interpret the periodic table to describe an element’s atomic makeup | Students will:-Interpret the periodic table to describe an element’s atomic makeup using their knowledge of the terms “isotope” and “ion” |
| **Body of Lesson****What activities and strategies will be used during the lesson?** | Observations and Experiments Lab | Unit Test | NotesLabDiscussion | NotesP/N/E ActivityWorksheet | Ions POGILIsotopes POGILNotes |
| **Reinforcement****What kind of follow-up assignments/homework will reinforce the lesson?** | STUDY | - | - | - | - |
| **Notes/Reminders** | \*Unit Test #1 Tuesday, 1/24/17\*P/N/E Quiz Friday, 1/27/17\*Isotopes and Ions Quiz Monday, 1/30/17 |